

## DEPARTMENTAL SEMINAR

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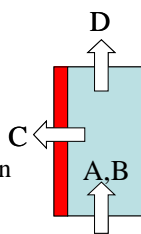


<b>TOPIC</b>	<b>Catalytic Membrane Reactors and Novel Trends in Catalysis</b>				
<b>SPEAKER</b>	<b>Prof J. Caro</b>				
<b>HOST</b>	<b>Prof Hong Liang</b>				
<b>DATE</b>	<b>27<sup>th</sup> February 2009 (Friday)</b>				
<b>TIME</b>	<b>10am</b>				
<b>VENUE</b>	<b>E5-02-32 , Faculty of Engineering, National University of Singapore</b> <a href="#">NUS Campus Map</a> & <a href="#">NUS: Faculty of Engineering</a>				
<b>SYNOPSIS</b>	<p><b>Catalytic Membrane Reactors</b></p> <p><i>Conversion enhancement in extractor type membrane reactors</i></p> <p>To overcome the equilibrium restriction for the thermodynamically controlled reactions, the reaction must be sufficient fast compared to the mass transport through the membrane (kinetic compatibility). The reactions in this category include dehydrogenations of alkanes to the corresponding olefins, esterification, steam reforming of methane to synthesis gas (CO, H<sub>2</sub>), Knoevenagel condensation, thermal water dissociation into H<sub>2</sub> and O<sub>2</sub>, and nitrous oxide (N<sub>2</sub>O) decomposition to N<sub>2</sub> and O<sub>2</sub>.</p> <p><i>Selectivity enhancement in distributor/contactor type membrane reactors</i></p> <p>In this case, the desired product is either an intermediate in a consecutive reaction or is one of the products in a system of parallel reactions. Reactions such as oxidation of hydrocarbons (including partial oxidations such as the oxidative dehydrogenation of alkanes to olefins and the oxidative coupling of methane to C<sub>2+</sub> hydrocarbons), the partial oxidation of methane to synthesis gas and partial hydrogenations of di- or multi-unsaturated hydrocarbons to less saturated ones belong to this category.</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="text-align: center;">Thermodynamically controlled reactions</th> <th style="text-align: center;">Kinetically controlled reactions</th> </tr> </thead> <tbody> <tr> <td style="text-align: center;"> <math>\Delta_R G^0</math> near zero  <math>\Delta_R G^0 = -RT \ln K \rightarrow K \approx 1</math>  <math>A + B \rightleftharpoons C + D</math> </td> <td style="text-align: center;"> <math>\Delta_R G^0</math> very negative  <math>\Delta_R G^0 = -RT \ln K \rightarrow K \gg 1</math>  <math>A + B \rightarrow C + D</math> </td> </tr> </tbody> </table>	Thermodynamically controlled reactions	Kinetically controlled reactions	$\Delta_R G^0$ near zero $\Delta_R G^0 = -RT \ln K \rightarrow K \approx 1$ $A + B \rightleftharpoons C + D$	$\Delta_R G^0$ very negative $\Delta_R G^0 = -RT \ln K \rightarrow K \gg 1$ $A + B \rightarrow C + D$
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### Extractor type membrane reactor

#### Conversion enhancement:

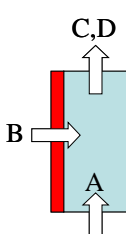
- Dehydrogenation
- Esterification
- Steam reforming
- Knoevenagel condensation
- Water splitting



### Distributor type membrane reactor

#### Selectivity enhancement:

- Hydrocarbon oxidation
- p-Xylene oxidation
- Methane to synthesis gas
- Partial hydrogenation
- Methane oxi-coupling

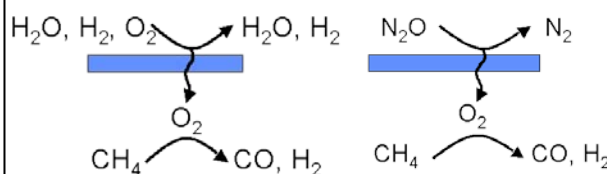


## Novel Trend in Catalysis

In this section, high throughput catalyst development by robots, parallel automatic testing of catalysts and catalysis in micro reactors will be presented.

### Membrane supported reactions

- Water splitting  
 $\text{H}_2\text{O} \rightleftharpoons \text{H}_2 + \frac{1}{2} \text{O}_2$  (left)
- Nitrous oxide abatement  
 $\text{N}_2\text{O} \rightarrow \text{N}_2 + \frac{1}{2} \text{O}_2$  (right)



## BIOGRAPHY



J. Caro is a full professor and the director of the Institute of Physical Chemistry and Electrochemistry at the University of Hannover since 2001. He served as the President of the German Catalysis Society in 2005 and 2006. He is also a member of the Board of Directors of the German Membrane Society and a speaker of the Lighthouse Project for the German Research Ministry (with 12 partners from industry and academic community). Prof Caro has published over 190 papers and is currently having 38 patents under his name. He is also a member of the Editorial Boards of 5 established journals, including *Advanced Materials*, *Micropor. Mesopor. Materials* and *Catalysis Communications*.

**ALL ARE WELCOME**

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